

Carbene-Mediated Transformations of 1-(Benzylideneamino)benzimidazoles

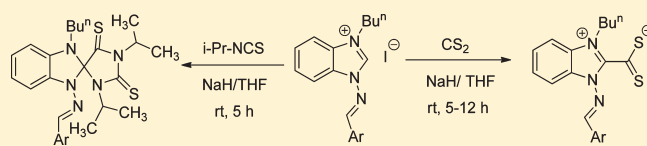
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S Supporting Information

ABSTRACT: Carbene-mediated transformations of *N*-(3-butylbenzimidazol-3-ium-1-yl)-1-arylmethanimine iodides with carbon disulfide and benzoyl isothiocyanate gave the corresponding NHC·CS₂ betaines in 68–85% and benzoyl-[1-butyl-3-[(*E*)-(aryl)methyleneamino]benzimidazol-1-ium-2-carbothioyl]azanides, respectively, in 74–85% yields. However, reaction with excess isopropyl isothiocyanate in NaH/THF at room temperature yielded the 1-butyl-1',3'-diisopropyl-3-[(*E*)-(aryl)methyleneamino]spiro[benzimidazole-2,5'-imidazolidine]-2',4'-dithiones (74–77%).



Carbenes derived from nitrogen heterocycles (NHC) are well-known organocatalysts for benzoin condensations,^{1,2} Stetter^{3,4} and Staudinger reactions,⁵ oxidative amidation,⁶ annulation of enals,⁷ ring-opening polymerizations,^{8,9} transesterifications,^{10,11} and other transformations;¹² they are also versatile ligands for transition metals.^{13–15} The σ -donating ability and strong nucleophilic character of NHC offers opportunities for the construction of substituted heterocycles^{16–21} and novel molecular frameworks.^{22,23}

An important property of NHCs is the ability to react with CO₂,^{24–26} CS₂,^{27–29} and isothiocyanates^{27,29} to form imidazol-2-ylidene, 1,3-thiazol-2-ylidene,^{17,30} 1,2,4-triazol-3-ylidene,^{17,31,32} and isothiazol-3-ylidene adducts.

A wide range of stable benzimidazolium or imidazol(in)ium–CS₂ adducts have been studied as novel ionic liquids,^{25,27,33} as catalysts in the cyanosilylation of aldehydes,³⁴ as intermediates for sulfur heterocycles,³⁵ ligands for gold complexes, surface units for gold nanoparticles,³⁶ and promising antifungal and antibacterial agents.^{29,37}

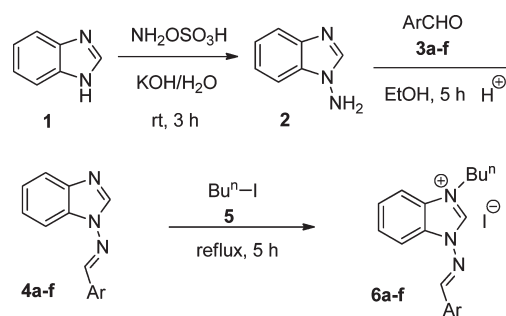
NHC zwitterionic betaine adducts with isothiocyanates (NHC·RNCS) are powerful intermediates for the synthesis of a variety of heterocycles.^{38,39} Thus, Cheng and co-workers utilized 2-thiocarbonyl benzimidazolium, imidazolium, and triazolium zwitterionic inner salts in [3 + 2] cycloaddition reactions, resulting in spiro-heterocycles.^{38,39}

Most literature NHC examples are based on imidazole and benzimidazole nuclei bearing various alkyl or aryl groups on nitrogen. An early report by Balch described a template synthesis of an *N*-amino-substituted *N*-heterocyclic carbene.⁴⁰ Lassaletta described the synthesis and application of a class of *N*-heterocyclic carbenes based on bis(*N,N*-dialkylamino)imidazolin-2-ylidines.^{41,42}

However, we found no literature reports of NHC·CS₂ or NHC·RNCS betaines or spirocyclic derivatives of *N*-(arylmethyleneamino)benzimidazoles. Herein, we describe our results from the carbene-mediated transformations of 1-(benzylideneamino)benzimidazoles.

1-Aminobenzimidazole (**2**) was synthesized from benzimidazole (**1**) and hydroxylamine-*O*-sulfonic acid in 75% yield following a literature procedure.⁴³ *N*-(Arylmethyleneimino)benzimidazoles **4a–f** were synthesized in 69–94% yield by the reaction of 1-aminobenzimidazole **2** with aldehydes **3a–f**, in ethanol in the presence of a catalytic quantity of sulfuric acid⁴⁴ (Scheme 1, Table 1). Quaternization of *N*-(arylmethyleneimino)benzimidazoles **4a–f** with butyl iodide **5** gave the corresponding 3-butyl-*N*-(arylmethyleneimino)benzimidazolium iodides **6a–f** in quantitative yields (Scheme 1, Table 1).

Scheme 1



For Ar: see Table 1

1,3-Dimethylbenzimidazolium iodide (**7**) was synthesized from benzimidazole following a literature procedure.⁴⁸ The singlet carbene generated in situ at the C-2 position of 1,3-dimethylbenzimidazolium iodide **7** by treatment with sodium hydride, on reaction with carbon disulfide **8**, gave 1,3-dimethylbenzimidazol-3-ium-2-carbodithioate (**10**) in 85% yield (Scheme 2). Deprotonation of

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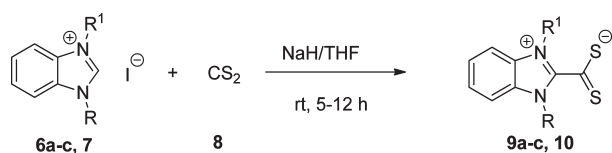
Table 1. Synthesis of *N*-(Arylmethyleneimino)benzimidazoles and Their Quaternized Salts

entry	Ar	products 4a–f		products 6a–f	
		yield (%)	mp (°C)	yield (%)	mp (°C)
a	4-CH ₃ O–C ₆ H ₄	92	93–95	100	154–155
b	C ₆ H ₅	94	125 ^a	100	171–172
c	4-Br-2-thiophenyl	69	174–175	100	175–176
d	4-CH ₃ –C ₆ H ₄	82	74–75	100	167–169
e	4-NO ₂ –C ₆ H ₄	79	225–226 ^b	100	217–219
f	4-Et ₃ N–C ₆ H ₄	91	118 ^c	100	151–152

^a Literature⁴⁵ mp 125–126 °C. ^b Literature⁴⁶ mp 223 °C. ^c Literature⁴⁷ mp not available.

N-(3-butylbenzimidazol-3-ium-1-yl)-1-(aryl)methanimine iodides **6a–c** using sodium hydride, followed by reaction with carbon disulfide **8** at room temperature, formed the corresponding NHC·CS₂ betaines (**9a–c**) (68–76%) (Scheme 2, Table 2). Isolation of the NHC·CS₂ betaine adducts was achieved by crystallization techniques in all cases. The structure of one representative example, **9a**, was confirmed by single-crystal X-ray diffraction (Figure 1).

More interesting results were obtained when the reaction of benzimidazolium salts with isopropyl isothiocyanate was carried out under similar experimental conditions. 1,3-Dimethylbenzimidazolium iodide (**7**) on reaction with excess of isopropyl isothiocyanate (**11**) in the presence of sodium hydride/THF at room temperature gave (1,3-dimethylbenzimidazol-3-ium-2-carbothioyl)-isopropylazanide (**12**) in 86% yield (Scheme 3).

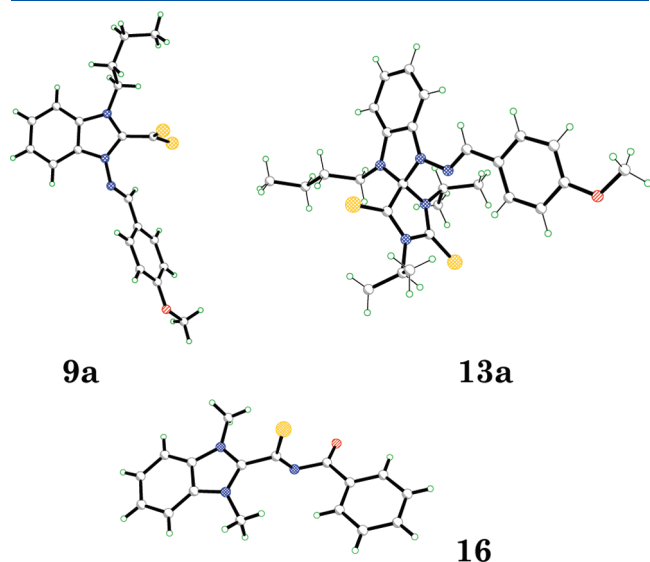
Scheme 2

For R and R¹: see Table 2

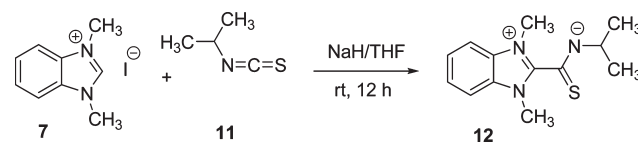
Table 2. Synthesis of NHC·CS₂ Betaine Adducts

reactant	product	R	R ¹	yield (%)	mp (°C)
6a	9a	–N=CH–(4-CH ₃ O–C ₆ H ₄)	Bu ⁿ	70	160–161
6b	9b	–N=CH–C ₆ H ₅	Bu ⁿ	76	171–172
6c	9c	–N=CH–(4-Br-2-thiophenyl)	Bu ⁿ	68	170–173
7	10	CH ₃	CH ₃	85	235–236 ^a

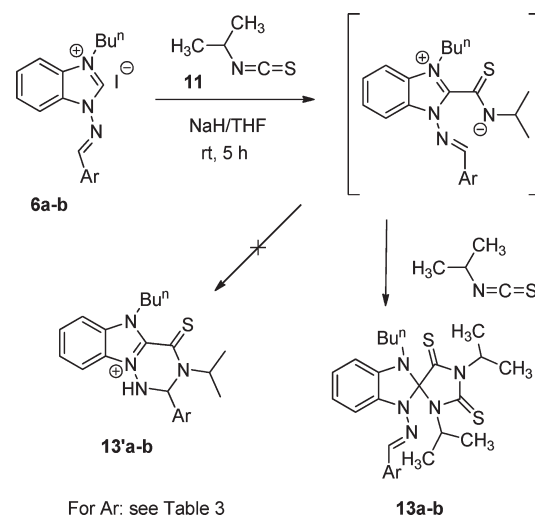
^a Literature³⁷ mp 237–238 °C.

**Figure 1.** X-ray structures of **9a**, **13a**, and **16**.

dazolium iodide (**7**) on reaction with excess of isopropyl isothiocyanate (**11**) in the presence of sodium hydride/THF at room temperature gave (1,3-dimethylbenzimidazol-3-ium-2-carbothioyl)-isopropylazanide (**12**) in 86% yield (Scheme 3).

Scheme 3

However, the reaction of *N*-(3-butylbenzimidazol-3-ium-1-yl)-1-(aryl)methanimine iodides (**6a,b**) with excess of isopropyl isothiocyanate **11** in the presence of NaH/THF at room temperature led to the formation of red crystalline compounds (**13a,b**) in 74–77% yield (Scheme 4, Table 3). The ¹H and ¹³C NMR spectra were complex and did not correspond to the formation of a betaine intermediate (Scheme 4). In fact, the spectroscopic data revealed two isopropyl groups, and the structure of representative example **13a** was confirmed by single-crystal X-ray diffraction studies (Figure 1). The X-ray crystallographic data proved the formation of 1-butyl-1',3'-diisopropyl-3-[(*E*)-(aryl)methyleneamino]spiro[benzimidazole-2,5'-imidazolidine]-2',4'-dithiones (**13a,b**), resulting from the addition of a second molecule of isopropyl isothiocyanate to the NHC·RNCS betaine adduct. In fact, with an imine function in the side chain, we were unable to isolate NHC·RNCS betaine adducts because the corresponding adducts reacted further with heteroallene even when the reagents were present in stoichiometric (1:1) or substoichiometric quantities. This result indicates that the intermolecular formation of **13a,b** is preferred over the intramolecular cyclization to **13'a,b** (Scheme 4). However, in the case of 1,3-dimethylbenzimidazolium iodide **7**, the reaction stopped at the NHC·RNCS betaine step and no spirocyclic product was observed even in presence of excess isopropyl isothiocyanate or at elevated temperatures.

Scheme 4

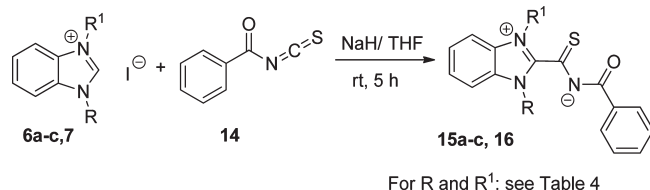
For Ar: see Table 3

Zwitterionic NHC·RNCS betaine intermediates were only isolated when a strongly deactivated isothiocyanate (benzoyl isothiocyanate) was used. Thus the reaction of benzimidazolium salts (**6a–c**, **7**) with equivalent amounts of benzoyl

Table 3. Synthesis of Spiro[imidazolidine-4,2'-indoline]-2,5-dithiones

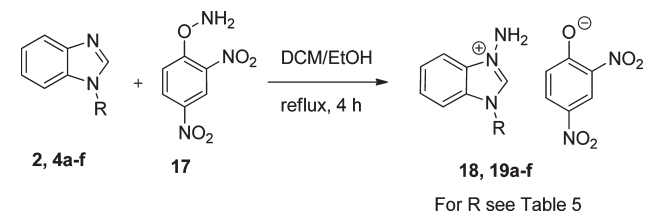
reactant	product	Ar	yield (%)	mp (°C)
6a	13a	4-CH ₃ O-C ₆ H ₄	77	141–142
6b	13b	C ₆ H ₅	74	116–119

isothiocyanate (**14**) in the presence of NaH/THF led to benzoyl benzimidazolium-2-carbonothioyl)amides (**15a–c**, **16**) in 74–85% yield (Scheme 5, Table 4). The structure of representative example **16** was confirmed by single-crystal X-ray diffraction studies (Figure 1).

Scheme 5**Table 4. Synthesis of NHC·RNCS Betaine Adducts**

reactant	product	R	R ¹	yield (%)	mp (°C)
6a	15a	-N=CH-(4-CH ₃ O-C ₆ H ₄)	Bu ⁿ	83	126–128
6b	15b	-N=CH-C ₆ H ₅	Bu ⁿ	77	125–127
6c	15c	-N=CH-(4-Br-2-thiophenyl)	Bu ⁿ	74	151–154
7	16	CH ₃	CH ₃	85	207–210

In an extension of this work, we studied the influence of functional groups on *N*-1 nitrogen, on the amination of 1-aminobenzimidazole and its derivatives. Compounds **2** and **4a–f** were reacted with 2,4-dinitrophenyl hydroxylamine (**17**) to form **18** and **19a–f** in 0–80% yield.⁴⁹ We found that imines with electron-donating substituents in the side chain gave higher yields compared to those with electron-withdrawing groups, and aminated products of some derivatives were not detected at all (Scheme 6, Table 5).

Scheme 6**Table 5. Amination of 1-Aminobenzimidazole and Its Imine Derivatives**

reactant	product	R	yield (%)	mp (°C)
2	18	-NH ₂	80	166–168
4a	19a	-N=CH-(4-CH ₃ O-C ₆ H ₄)	42	146–148
4b	19b	-N=CH-C ₆ H ₅	0	
4c	19c	-N=CH-(4-Br-2-thiophenyl)	0	
4d	19d	-N=CH-(4-CH ₃ -C ₆ H ₄)	40	132–134
4e	19e	-N=CH-(4-NO ₂ -C ₆ H ₄)	0	
4f	19f	-N=CH-(4-Et ₃ N-C ₆ H ₄)	60	180

EXPERIMENTAL SECTION

Melting points were determined on a capillary point apparatus equipped with a digital thermometer and are uncorrected. NMR spectra were recorded in CDCl₃ or DMSO-*d*₆ with TMS for ¹H (300 MHz) and ¹³C (75 MHz) as an internal reference. Aminating agent **17** was prepared via a two-step literature procedure.⁵⁰ Elemental analysis was performed on a Carlo Erba-1106 instrument.

General Method for the Preparation of Imines 4a–f: 1-Aminobenzimidazole and corresponding aldehyde (1.05 equiv) were dissolved in absolute ethanol and stirred in the presence of a catalytic amount of sulfuric acid. After 5 h, the solvent was removed under reduced pressure and the imines were obtained by recrystallization from EtOAc/hexanes.

***N*-(Benzimidazol-1-yl)-1-(4-methoxyphenyl)methanimine (4a):** White microcrystals (92%), mp 93–95.0 °C; ¹H NMR (CDCl₃) δ 8.69 (s, 1H), 8.32 (s, 1H), 7.85–7.78 (m, 3H), 7.75–7.70 (m, 1H), 7.39–7.28 (m, 2H), 7.02–6.97 (m, 2H), 3.87 (s, 3H); ¹³C NMR (CDCl₃) δ 162.5, 152.2, 142.5, 136.5, 131.9, 129.9, 125.6, 123.9, 123.0, 120.7, 114.5, 110.8, 55.5. Anal. Calcd for C₁₅H₁₃N₃O: C, 71.70; H, 5.21; N, 16.72. Found: C, 71.87; H, 5.28; N, 16.78.

***N*-(Benzimidazol-1-yl)-1-phenylmethanimine (4b):** White microcrystals (94%), mp 125.0 °C (lit.⁴⁵ mp 125–126 °C); ¹H NMR (CDCl₃) δ 8.76 (d, *J* = 2.7 Hz, 1H), 8.37 (s, 1H), 7.91–7.86 (m, 2H), 7.84–7.80 (m, 1H), 7.78–7.73 (m, 1H), 7.52–7.47 (m, 3H), 7.41–7.29 (m, 2H); ¹³C NMR (CDCl₃) δ 151.7, 142.5, 136.6, 132.9, 131.8, 131.6, 129.0, 128.1, 124.1, 123.1, 120.7, 110.8. Anal. Calcd for C₁₄H₁₁N₃: C, 76.00; H, 5.01; N, 18.99. Found: C, 75.82; H, 4.99; N, 19.07.

***N*-(Benzimidazol-1-yl)-1-(4-bromo-2-thienyl)methanimine (4c):** Colorless crystals (79%), mp 174.0–175.0 °C; ¹H NMR (DMSO-*d*₆) δ 9.36 (s, 1H), 8.90 (s, 1H), 7.99 (s, 1H), 7.78 (d, *J* = 7.8 Hz, 1H), 7.74 (d, *J* = 7.8 Hz, 1H), 7.69 (s, 1H), 7.39 (t, *J* = 7.5 Hz, 1H), 7.32 (t, *J* = 7.5 Hz, 1H); ¹³C NMR (DMSO-*d*₆) δ 146.4, 141.6, 138.9, 137.4, 134.4, 131.4, 128.1, 123.8, 122.8, 119.9, 110.6, 109.6. Anal. Calcd for C₁₂H₈BrN₃S: C, 47.07; H, 2.63; N, 13.72. Found: C, 47.12; H, 2.59; N, 13.60.

***N*-(Benzimidazol-1-yl)-1-(*p*-tolyl)methanimine (4d):** White microcrystals (82%), mp 74–75.0 °C; ¹H NMR (CDCl₃) δ 8.71 (s, 1H), 8.38 (s, 1H), 7.83–7.72 (m, 4H), 7.39–7.25 (m, 4H), 2.42 (s, 3H); ¹³C NMR (CDCl₃) δ 152.4, 142.4, 142.3, 136.6, 131.9, 130.3, 129.8, 128.2, 124.1, 123.2, 120.7, 110.9, 21.7. Anal. Calcd for C₁₅H₁₃N₃: C, 76.57; H, 5.57; N, 17.86. Found: C, 76.67; H, 5.61; N, 17.99.

***N*-(Benzimidazol-1-yl)-1-(4-nitrophenyl)methanimine (4e):** Yellow microcrystals (79%), mp 225–226 °C (lit.⁴⁶ mp 223 °C); ¹H NMR (DMSO-*d*₆) δ 9.36 (s, 1H), 9.03 (s, 1H), 8.40 (d, *J* = 7.8 Hz, 2H), 8.20 (d, *J* = 7.7 Hz, 2H), 7.90 (d, *J* = 7.8 Hz, 1H), 7.75 (d, *J* = 7.8 Hz, 1H), 7.42 (t, *J* = 7.5 Hz, 1H), 7.34 (t, *J* = 7.5 Hz, 1H); ¹³C NMR (DMSO-*d*₆) δ 150.1, 148.5, 141.7, 139.2, 137.1, 131.8, 128.8, 124.1, 123.0, 120.0, 110.8. Anal. Calcd for C₁₄H₁₀N₄O₂: C, 63.15; H, 3.79; N, 21.04. Found: C, 63.03; H, 3.70; N, 20.79.

4-[(*E*)-Benzimidazol-1-yliminomethyl]-*N,N*-diethylaniline (4f): Green crystals (91%), mp 118.0 °C (lit.⁴⁷ mp not available); ¹H NMR (DMSO-*d*₆) δ 8.96 (s, 1H), 8.89 (s, 1H), 7.77–7.70 (m, 4H), 7.35 (t, *J* = 7.4 Hz, 1H), 7.27 (t, *J* = 7.6 Hz, 1H), 6.78 (d, *J* = 9.0 Hz, 2H), 3.42 (q, *J* = 7.0 Hz, 4H), 1.13 (t, *J* = 6.9 Hz, 6H); ¹³C NMR (DMSO-*d*₆) δ 154.6, 149.9, 141.5, 136.5, 132.0, 130.0, 123.2, 122.1, 119.7, 119.1, 111.0, 110.5, 43.7, 12.3. Anal. Calcd for C₁₈H₂₀N₄: C, 73.94; H, 6.89; N, 19.16. Found: C, 73.56; H, 7.00; N, 18.99.

General Method for the Preparation of *N*-(3-Butylbenzimidazol-3-ium-1-yl)-1-arylmethanimine iodides (6a–f): 1-Iodobutane (3 equiv) and the corresponding imines **4a–f** were heated with stirring at 100 °C in a round-bottom flask for 5 h without solvent. Excess 1-iodobutane was removed under reduced pressure at 50 °C to give the corresponding 3-butyl-*N*-(arylmethyleneimino)benzimidazolium iodides **6a–f** in quantitative yields.

***N*-(3-Butylbenzimidazol-3-ium-1-yl)-1-(4-methoxyphenyl)methanimine iodide (6a):** White crystals (100%), mp 154.0–155.0 °C;

(d, $J = 7.0$ Hz, 3H), 0.86 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (DMSO- d_6) δ 139.0, 136.6, 134.8, 131.9, 129.2, 128.7, 125.8, 121.8, 118.2, 108.3, 105.0, 103.6, 42.8, 29.5, 19.8, 19.0, 17.9, 13.6. Anal. Calcd for $\text{C}_{26}\text{H}_{33}\text{N}_5\text{S}_2$: C, 65.10; H, 6.93; N, 14.60. Found: C, 65.10; H, 7.06; N, 14.58.

General Method for the Preparation of NHC-RC(O)NCS Betaines 15a–c, 16. Sodium hydride (1.2 equiv) was added to a mixture of benzoylisothiocyanate and the corresponding *N*-(3-butylbenzimidazol-3-ium-1-yl)-1-(aryl)methanimine iodides **6a–c** in 1/1 molar ratio in dry THF and stirred for 5 h (12 h in the case of **7**) at room temperature under argon. The solution was filtered, the filtrate concentrated under reduced pressure, and the residue was purified by flash chromatography using EtOAc/hexanes to obtain betaine adducts **15a–c**, **16**.

Benzoyl-[1-butyl-3-[(*E*)-(4-methoxyphenyl)methyleneamino]benzimidazol-1-ium-2-carbothiopyl]azanide (15a): Yellow microcrystals (83%), mp 126.0–128.0 °C; ^1H NMR (DMSO- d_6) δ 9.35 (s, 1H), 8.13 (d, $J = 8.1$ Hz, 1H), 8.03–7.94 (m, 5H), 7.75–7.65 (m, 2H), 7.53 (t, $J = 7.4$ Hz, 1H), 7.34 (t, $J = 7.7$ Hz, 2H), 7.18 (d, $J = 8.7$ Hz, 2H), 4.63 (t, $J = 7.2$ Hz, 2H), 3.89 (s, 3H), 2.02–1.92 (m, 2H), 1.47–1.36 (m, 2H), 0.91 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (DMSO- d_6) δ 179.1, 169.8, 168.2, 163.6, 145.1, 133.7, 132.3, 131.6, 129.2, 128.9, 128.3, 127.5, 126.8, 123.8, 114.8, 113.6, 113.2, 55.7, 45.2, 30.6, 19.4, 13.5. Anal. Calcd for $\text{C}_{27}\text{H}_{26}\text{N}_4\text{O}_2\text{S}$: C, 68.91; H, 5.57; N, 11.91. Found: C, 68.63; H, 5.63; N, 11.73.

Benzoyl-[1-[(*E*)-benzylideneamino]-3-butylbenzimidazol-3-ium-2-carbothiopyl]azanide (15b): Yellow crystals (77%), mp 125.0–127.0 °C; ^1H NMR (DMSO- d_6) δ 9.48 (s, 1H), 8.18–8.08 (m, 2H), 8.04–7.94 (m, 4H), 7.78–7.69 (m, 3H), 7.63 (t, $J = 7.4$ Hz, 2H), 7.53 (t, $J = 7.4$ Hz, 1H), 7.34 (t, $J = 7.5$ Hz, 2H), 4.64 (t, $J = 7.3$ Hz, 2H), 2.05–1.93 (m, 2H), 1.43 (sextet, $J = 7.4$ Hz, 2H), 0.92 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (DMSO- d_6) δ 179.0, 169.8, 168.1, 145.3, 133.7, 133.5, 132.3, 131.3, 129.3, 129.1, 128.9, 128.2, 127.2, 126.9, 126.9, 113.6, 113.2, 45.2, 30.5, 19.3, 13.4. Anal. Calcd for $\text{C}_{26}\text{H}_{24}\text{N}_4\text{OS}$: C, 70.88; H, 5.49; N, 12.72. Found: C, 70.57; H, 5.60; N, 12.51.

Benzoyl-[1-[(*E*)-(4-bromo-2-thienyl)methyleneamino]-3-butylbenzimidazol-3-ium-2-carbothiopyl]azanide (15c): Yellow crystals (74%), mp 151.0–154.0 °C; ^1H NMR (DMSO- d_6) δ 9.55 (s, 1H), 8.18 (s, 1H), 8.11 (d, $J = 8.1$ Hz, 1H), 8.05 (d, $J = 7.3$ Hz, 1H), 7.98 (d, $J = 7.2$ Hz, 2H), 7.87 (s, 1H), 7.75–7.66 (m, 2H), 7.55 (t, $J = 7.2$ Hz, 1H), 7.41 (t, $J = 7.4$ Hz, 2H), 4.61 (t, $J = 7.2$ Hz, 2H), 1.99–1.90 (m, 2H), 1.46–1.34 (m, 2H), 0.89 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (DMSO- d_6) δ 179.1, 170.1, 161.0, 145.6, 138.7, 136.5, 133.8, 132.5, 131.8, 129.4, 129.0, 128.6, 128.4, 127.1, 127.1, 113.8, 113.3, 110.4, 45.4, 30.5, 19.4, 13.5. Anal. Calcd for $\text{C}_{24}\text{H}_{21}\text{BrN}_4\text{OS}_2$: C, 54.86; H, 4.03; N, 10.66. Found: C, 55.19; H, 4.09; N, 10.42.

Benzoyl-(1,3-dimethylbenzimidazol-3-ium-2-carbothiopyl)azanide (16): Yellow crystals (85%), mp 207.0–210.0 °C; ^1H NMR (DMSO- d_6) δ 8.02–7.94 (m, 4H), 7.69–7.64 (m, 2H), 7.59 (d, $J = 6.7$ Hz, 1H), 7.55–7.48 (m, 2H), 4.02 (s, 6H); ^{13}C NMR (DMSO- d_6) δ 178.7, 172.0, 150.0, 134.2, 132.4, 130.4, 129.2, 128.5, 126.2, 113.1, 31.7.

General Method for Amination. To a solution of 1-aminobenzimidazole **2** or its imine derivatives **4a–f** in dichloromethane/ethanol (5:1) was added 1.2–1.5 equiv of *O*-(2,4-dinitrophenyl)hydroxylamine with stirring over a period of 2–3 min. The mixture was heated under reflux for 5–8 h. After cooling and standing for 30 min, the solid was filtered off, washed twice with ethyl acetate (20 mL), and dried under vacuum to obtain the corresponding 2,4-dinitrophenolate salts.

Benzimidazol-3-ium-1,3-diamine; 2,4-Dinitrophenolate (18): Brown microcrystals (80%), mp 166.0–168.0 °C; ^1H NMR (DMSO- d_6) δ 9.88 (s, 1H), 8.60 (d, $J = 3.0$ Hz, 1H), 7.93–7.87 (m, 2H), 7.80 (dd, $J = 9.8$, 3.2 Hz, 1H), 7.75–7.67 (m, 2H), 6.91 (s, 4H), 6.34 (d, $J = 9.9$ Hz, 1H); ^{13}C NMR (DMSO- d_6) δ 169.7, 142.4, 136.1, 131.0, 127.8, 127.5, 126.5, 126.1, 124.8, 112.9. Anal. Calcd for $\text{C}_{13}\text{H}_{12}\text{N}_6\text{O}_5$: C, 46.99; H, 3.64; N, 25.29. Found: C, 46.96; H, 3.49; N, 25.14.

3-[(*E*)-(4-Methoxyphenyl)methyleneamino]benzimidazol-1-ium-1-amine; 2,4-Dinitrophenolate (19a): Yellow microcrystals (42%), mp 146.0–148.0 °C; ^1H NMR (DMSO- d_6) δ 10.40 (s, 1H), 9.23 (s, 1H), 8.58 (d, $J = 3.9$ Hz, 1H), 8.13–8.09 (m, 1H), 8.01–7.95 (m, 3H), 7.80–7.72 (m, 3H), 7.19 (d, $J = 8.8$ Hz, 2H), 7.13 (s, 2H), 6.30 (d, $J = 9.8$ Hz, 1H); ^{13}C NMR (DMSO- d_6) δ 170.3, 165.1, 163.4, 162.5, 136.0, 131.3, 131.1, 128.5, 127.4, 126.9, 126.5, 125.0, 123.7, 114.9, 113.3, 112.9, 109.2, 55.7. Anal. Calcd for $\text{C}_{21}\text{H}_{18}\text{N}_6\text{O}_6$: C, 56.00; H, 4.03; N, 18.66. Found: C, 55.86; H, 3.86; N, 18.58.

3-[(*E*)-*p*-Tolylmethyleneamino]benzimidazol-1-ium-1-amine; 2,4-Dinitrophenolate (19d): Brown microcrystals (40%), mp 132.0–134.0 °C; ^1H NMR (DMSO- d_6) δ 10.45 (s, 1H), 9.28 (s, 1H), 8.58 (d, $J = 3.3$ Hz, 1H), 8.14–8.11 (m, 1H), 8.01–7.97 (m, 1H), 7.92 (d, $J = 7.9$ Hz, 2H), 7.80–7.72 (m, 3H), 7.45 (d, $J = 7.9$ Hz, 2H), 7.16 (s, 2H), 6.32 (d, $J = 9.8$ Hz, 1H), 2.43 (s, 3H); ^{13}C NMR (DMSO- d_6) δ 169.9, 162.5, 143.9, 136.0, 135.9, 131.0, 129.9, 129.1, 128.6, 128.5, 127.7, 127.5, 127.4, 126.9, 126.3, 124.8, 113.2, 112.9, 21.3. Anal. Calcd for $\text{C}_{21}\text{H}_{18}\text{N}_6\text{O}_5$: C, 58.06; H, 4.18; N, 19.35. Found: C, 57.64; H, 4.05; N, 19.38.

3-[(*E*)-[4-(Diethylamino)phenyl]methyleneamino]benzimidazol-1-ium-1-amine; 2,4-Dinitrophenolate (19f): Brown microcrystals (60%), mp 180.0 °C; ^1H NMR (DMSO- d_6) δ 10.38 (s, 1H), 9.01 (s, 1H), 8.63 (d, $J = 2.7$ Hz, 1H), 8.04 (d, $J = 7.5$ Hz, 1H), 7.98 (d, $J = 7.8$ Hz, 1H), 7.88 (dd, $J = 9.6$, 3.0 Hz, 1H), 7.80–7.68 (m, 4H), 7.09 (br s, 2H), 6.83 (d, $J = 8.7$ Hz, 2H), 6.50 (d, $J = 9.6$ Hz, 1H), 3.46 (q, $J = 6.6$ Hz, 4H), 1.16 (t, $J = 6.8$ Hz, 6H); ^{13}C NMR (DMSO- d_6) δ 168.6, 162.8, 151.2, 136.0, 135.7, 131.5, 131.1, 129.0, 128.5, 127.7, 127.0, 126.7, 125.5, 124.5, 117.0, 113.1, 112.8, 111.1, 43.9, 12.3. Anal. Calcd for $\text{C}_{24}\text{H}_{25}\text{N}_7\text{O}_5$: C, 58.65; H, 5.13; N, 19.95. Found: C, 58.32; H, 5.26; N, 19.88.

ASSOCIATED CONTENT

Supporting Information. ^1H and ^{13}C spectra for **4a–f**, **6a–f**, **7**, **9a–c**, **10**, **12**, **13a,b**, **15a–c**, **16**, **18**, **19a**, **19d**, and **19f** and the X-ray crystal structure and data of **9a**, **13a**, and **16** as CIF files. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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